

Chromatographic Methods of Analysis

Section: 7 HPLC

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Liquid chromatography

Applied to the separation of any compound that is soluble in a liquid phase; biological compounds, synthetic or natural polymers, and inorganic compounds.

- Use large, non-rigid support material, particles size $> 150 \mu\text{m}$.
- Column; diameter 10 ~ 50 mm, column length L: 50 ~ 500 cm, flow rate F: $< 1 \text{ ml /min}$.
- Gravity; Large H, small N.
- Poor system efficiencies and large plate heights.

Basic Principles of HPLC

- **HPLC High Performance Liquid Chromatography**
(Also called; High Pressure Liquid Chromatography)
- It is a form of liquid column chromatography used for separation, identification, and quantification of compounds in bio- and analytical chemistry.
- It represents, adsorption, ion exchange and gel chromatography in the form of column.
- It is really the automation of traditional liquid chromatography under conditions which provide enhanced resolution separations during shorter periods of time.

Types of HPLC Separations

■ Normal Phase:

Separation of polar components by using a polar bonded stationary phase (like silica or alumina) and non-polar solvents (like hexane).

■ Reversed Phase:

Separation of non-polar analytes by using a non-polar bonded stationary phase (like polystyrene) and a polar solvents (like methanol).

Property	Normal	Reversed-Phase
Packing Polarity	Polar	Non-Polar
Mobile Phase Polarity	Non-Polar	Polar
Elution Order	Non-Polar first	Polar First first
Increasing Polarity of Mobile Phase	Reduces t_R	Increases t_R

Requirements for HPLC

■ Column:

- Stainless steel tubing for high pressure, or heavy-wall glass for low P (< 600 psi).
- Analytical column: straight with length (5 ~ 25 cm), diameter (3 ~ 5 mm), particles size (3 ~ 5 μm).

■ Temperature control: < 150 °C \pm 0.1 °C.

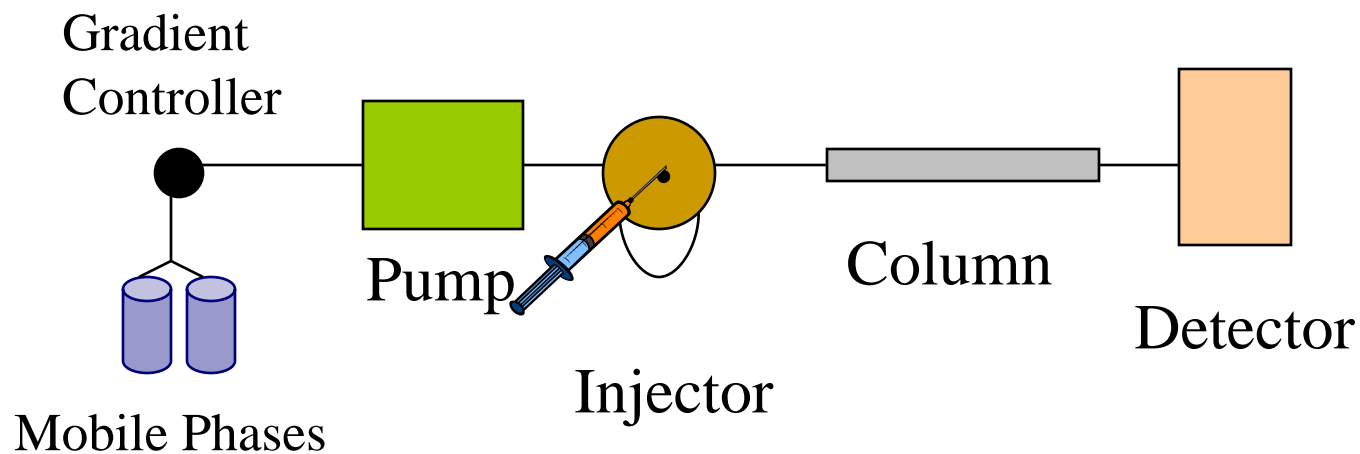
■ Column packing: silica, alumina, a polystyrene-divinylbenzene synthetic or an ion-exchange resin.

■ Solvent Reservoirs (can be a single solvent or multi-solvents).

■ Apply high pressure to force liquid through the beads faster (pressures up to 100 atm)

■ Control flow rate (from 0.1 to 10 ml/min).

Instrumentation



HPLC Chromatograph

Hewlett-Packard Series 1100 HPLC

solvent →

pump →

injector →

column →

detector →





Detectors

- **UV detectors:**
 - **Single wavelength (filter).**
 - **Variable wavelength (monochromator).**
 - **Multiple wavelengths.**
- **Fluorescence**
- **Electrochemical (Amperometry, voltammetry, coulometry and conductometry).**
- **Mass Spectrometric.**
- **Refractive index detector.**



■ Advantages of HPLC

- High speed (separation in few minutes).
- High resolution.
- High sensitivity (separate samples up to ng or fg).
- Reproducibility of $\pm 1\%$ (not so for LC).
- High accuracy.
- Automation.
- Separation of large number of components (upto 20 component can be separated).

■ Disadvantages of HPLC

- High cost.
- Complexity.
- Low sensitivity for some compounds.
- Irreversibly adsorbed compounds can not be detected.
- Co-elution difficult to be detect.

Applications of HPLC

- Separation of typical non-volatile compounds as:
 - Pharmaceuticals like aspirin, ibuprofen, or acetaminophen.
 - Salts like sodium chloride and potassium phosphate.
 - Proteins like egg white or blood protein.
 - Organic chemicals like polymers (e.g. polystyrene, polyethylene).
 - Heavy hydrocarbons like asphalt or motor oil.
 - Many natural products such as ginseng, herbal medicines, plant extracts.
- Separation of thermally unstable compounds as:
 - Trinitrotoluene (TNT).
 - Enzymes .
 - Polychlorinated hydrocarbons.